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## Key indicators

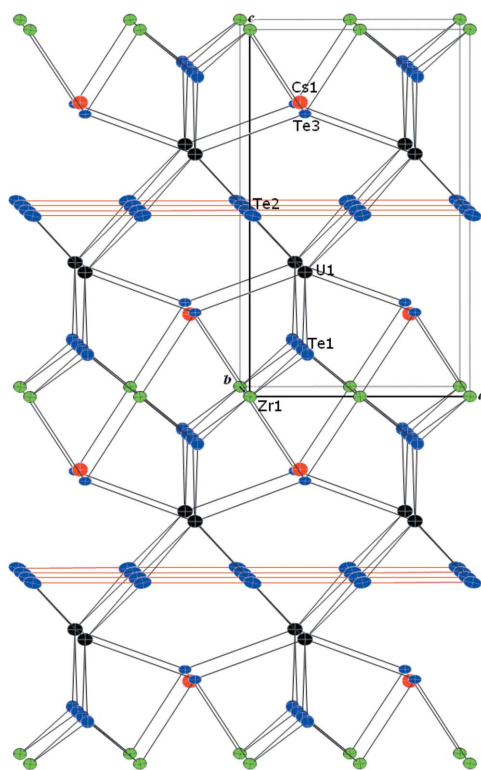
Single-crystal X-ray study  
 $T = 153\text{ K}$   
Mean  $\sigma(\text{Zr}-\text{Te}) = 0.005\text{ \AA}$   
 $R$  factor = 0.025  
 $wR$  factor = 0.067  
Data-to-parameter ratio = 28.6For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.Caesium zirconium uranium pentatelluride,  
 $\text{CsZrUTe}_5$ 

$\text{CsZrUTe}_5$  is isostructural with  $\text{CsTiUTe}_5$ . In the asymmetric unit, the site symmetries of the Cs, U, Zr, Te1, Te2 and Te3 atoms are  $mm2$ ,  $mm2$ ,  $2/m..$ ,  $m..$ ,  $2.$  and  $mm2$ , respectively.  $\text{CsZrUTe}_5$  has a layered structure that contains  $\text{UTe}_8$  bicapped trigonal prisms sharing a common edge with  $\text{ZrTe}_6$  octahedra. Cs cations separate the layers. The structure contains an infinite linear Te–Te chain, with Te atoms separated by  $3.1551(4)\text{ \AA}$ .

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## Comment

The structure of the compound  $\text{CsTiUTe}_5$  (Cody & Ibers, 1995) features an infinite linear chain of short Te–Te interactions, the Te–Te distance being  $3.065(1)\text{ \AA}$ . Despite the presence of this chain, the compound is a semiconductor. Crystals sufficiently large to allow more complete conductivity measurements could not be grown. Accordingly, we turned our attention to the synthesis of the possible Zr analogue, namely  $\text{CsZrUTe}_5$ . The compound has been synthesized and its structure is reported here. Unfortunately, large crystals of this material have not been grown.



**Figure 1**  
The structure of  $\text{CsZrUTe}_5$ , viewed approximately along  $[010]$ . The Te–Te bonds in the infinite chain are drawn as red lines. Displacement ellipsoids are displayed at the 99% probability level.

Fig. 1 shows the crystal structure of CsZrUTe<sub>5</sub>, which is isostructural with CsTiUTe<sub>5</sub> (Cody & Ibers, 1995). The anionic layers comprise UTe<sub>8</sub> bicapped trigonal prisms and ZrTe<sub>6</sub> octahedra. These layers are separated by Cs cations which are coordinated in a bicapped pentagonal prism of Te atoms.

The U—Te distances and U-atom coordination environment are similar to those found in UTe<sub>2</sub> (Beck & Dausch, 1988). In both structures, the UTe<sub>8</sub> bicapped trigonal prisms have a short Te—Te distance at one edge of the triangular face. In CsZrUTe<sub>5</sub> the prisms share edges of a rectangular face through Te<sub>2</sub> atoms in the [100] direction. This results in an infinite linear chain of Te atoms; the Te—Te distance is 3.1551 (4) Å.

Each nearly regular ZrTe<sub>6</sub> octahedron shares faces with two adjacent octahedra to create an unusual infinite [ZrTe<sub>3</sub><sup>2-</sup>] chain. The Zr—Te distances (Table 1) are comparable with those of 2.8771 (7)–3.0426 (8) Å in ZrTe (Örlygsson & Harbrecht, 2001).

## Experimental

CsZrUTe<sub>5</sub> was obtained as black needles from the reaction of Cs<sub>2</sub>Te<sub>3</sub> (0.11 mmol), U (Oak Ridge National Laboratory, 0.23 mmol), Zr (Chemtall GmbH, 0.23 mmol) and Te (Aldrich, 99.8%, 0.80 mmol) with CsCl (Strem, 99.999%) as flux. The Cs<sub>2</sub>Te<sub>3</sub> reactive flux was prepared by the stoichiometric reaction of Cs (Alfa Aesar, 99.8%) and Te in liquid NH<sub>3</sub> at 194 K. The reactants were loaded into a fused-silica tube under an Ar atmosphere in a dry box. The tube was evacuated to 10<sup>-4</sup> Torr (1 Torr = 133.322 Pa), sealed and placed in a computer-controlled furnace. The sample was heated to 1223 K, kept at 1223 K for 144 h and then slowly cooled at 2 K h<sup>-1</sup> to 293 K. The resulting crystals were extracted manually from the melt.

### Crystal data

CsZrUTe <sub>5</sub>	Z = 2
<i>M<sub>r</sub></i> = 1100.16	<i>D<sub>x</sub></i> = 6.675 Mg m <sup>-3</sup>
Orthorhombic, <i>Pm</i> m	Mo <i>K</i> α radiation
<i>a</i> = 6.3101 (8) Å	<i>μ</i> = 31.99 mm <sup>-1</sup>
<i>b</i> = 8.2299 (10) Å	<i>T</i> = 153 (2) K
<i>c</i> = 10.5401 (13) Å	Needle, black
<i>V</i> = 547.36 (12) Å <sup>3</sup>	0.336 × 0.062 × 0.056 mm

### Data collection

Bruker SMART 1000 CCD area-detector diffractometer	6537 measured reflections
<i>ω</i> scans	801 independent reflections
Absorption correction: numerical face-indexed ( <i>SHELXTL</i> ; Sheldrick, 2003)	769 reflections with <i>I</i> > 2σ( <i>I</i> )
<i>T<sub>min</sub></i> = 0.039, <i>T<sub>max</sub></i> = 0.217	<i>R<sub>int</sub></i> = 0.038
	<i>θ<sub>max</sub></i> = 28.9°

### Refinement

Refinement on <i>F</i> <sup>2</sup>	$w = 1/[\sigma^2(F_o^2) + (0.04P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.025$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.067$	( $\Delta/\sigma$ ) <sub>max</sub> < 0.001
<i>S</i> = 1.27	$\Delta\rho_{\text{max}} = 3.85 \text{ e } \text{Å}^{-3}$
801 reflections	$\Delta\rho_{\text{min}} = -2.55 \text{ e } \text{Å}^{-3}$
28 parameters	

**Table 1**

Selected bond lengths (Å).

U1—Te1	3.0961 (6)	Te2—Te2 <sup>ii</sup>	3.1551 (4)
U1—Te2	3.1118 (4)	Cs1—Te1 <sup>iii</sup>	3.8725 (5)
U1—Te3 <sup>i</sup>	3.3599 (4)	Cs1—Te2	3.9828 (6)
Zr1—Te3 <sup>i</sup>	2.8890 (6)	Cs1—Te1 <sup>iv</sup>	4.0608 (8)
Zr1—Te1	2.9087 (4)	Cs1—Te3	4.1193 (5)

Symmetry codes: (i)  $-x, -y, -z + 1$ ; (ii)  $x + \frac{1}{2}, y, -z + 1$ ; (iii)  $-x, -y + 1, -z + 1$ ; (iv)  $x, y, z + 1$ .

The structure was standardized by means of the program *STRUCTURE TIDY* (Gelato & Parthé, 1987). The highest peak is 0.03 Å and the deepest hole is 1.82 Å from atom U.

Data collection: *SMART* (Bruker, 2003); cell refinement: *SMART*; data reduction: *SAINTE-Plus* (Bruker, 2003); program(s) used to solve structure: *SHELXTL* (Sheldrick, 2003); program(s) used to refine structure: *SHELXTL*; molecular graphics: *CRYSTALMAKER* (*CRYSTALMAKER* Software, 2005); software used to prepare material for publication: *SHELXTL*.

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